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Review: Acids, Bases, and pH 1. Know the properties of acids and bases. What are their similarities and differences? Acids Taste sour React with metals to form hydrogen gas Contain a hydrogen ion (H⁺) pH value less than 7 Bases Exam #10 Review: Acids, Bases, and pH TEST TOMORROW! Make sure to memorize the 7 strong acids! Know how to find pH, pOH, H⁺, OH⁻, K_a, K_b (this review is based on Woods' key) USE SUBSCRIPTS IN T...Chemistry- Acids and Bases Test Review Flashcards | Quizlet CHAPTER 14 REVIEW . Acids and Bases. SHORT ANSWER Answer the following questions in the space provided. 1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1: H₂SO₃(aq) + H₂O(l) ~ H₃O⁺(aq) + HSO₃⁻(aq) b. CHAPTER 14 REVIEW Acids and Bases - Weebly Naming Acids and Bases Answers Name the following acids and bases: 1) NaOH sodium hydroxide 2) H₂SO₃ sulfurous acid 3) H₂S hydrosulfuric acid 4) H₃PO₄ phosphoric acid 5) NH₃ ammonia 6) HCN hydrocyanic acid 7) Ca(OH)₂ calcium hydroxide 8) Fe(OH)₃ iron (III) hydroxide 9) H₃PO₄ hydrophosphoric acid Write the formulas of the following acids ...Naming Acids and Bases - Ms. Mogck's Classroom Start studying Chemistry Chapter 14 ~Acids and Bases Test Review. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Chemistry Chapter 14 ~Acids and Bases Test Review ...Acid Base Test Prep. ... Students will review material about acids and bases using a practice test and a Jeopardy game. Big Idea. Acids and bases is a broad topic that includes topics like pH, neutralization reactions, and titrations. ... Several students

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HSO₄⁻ is A)H₂SO₄ B)SO₄²⁻ C)H₃SO₄⁺ D)HSO₄⁺ E)OH⁻ 2) The conjugate acid of ...

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H₂SO₃ 3) H₂S hydrosulfuric acid 4) H₃PO₄ 4) phosphoric acid 5) NH₃ ammonia 6)

HCN hydrocyanic acid 7) Ca(OH)₂ calcium hydroxide 8) Fe(OH)₃ iron (III) hydroxide 9) H₃P

hydrophosphoric acid Write the formulas of the following acids ...

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